SOLVED PAPER	NEET (UG) 07th May 2023	Code F3

Important Instructions

1. *The test is of **3 hours 20 minutes** duration and Test Booklet contains **200** multiple choice questions (four options with a single correct answer) from **Physics, Chemistry and Biology (Botany and Zoology)**. **50** Questions in each subject are divided into two **Section (A and B)** as per details given below:*
   - *(a) **Section A** shall consist of **35 (Thirty-five)** Questions in each subject (Question Nos- 1 to 35, 51 to 85, 101 to 135 and 151 to 185). All questions are compulsory.*
   - *(b) **Section B** shall consist of **15 (Fifteen)** Questions in each subject (Question Nos- 36 to 50, 86 to 100, 136 to 150 and 80 to 200). In Section B, a candidate needs to **attempt any 10 (Ten)** questions out of **15 (Fifteen)** in each subject.*

   Candidates are advised to read all 15 questions in each subject of Section B** before they start attempting the question paper. In the event of a candidate attempting more than ten questions, **the first ten questions answered by the candidate shall be evaluated.
2. *Each question carries **4** marks. For each correct response, the candidate will get **4** marks. For each incorrect response, one mark will be deducted from the total scores. **The maximum marks are 720.***
3. *Use **Blue/Black Ball Point Pen only** for writing particulars on this page/marking responses on Answer Sheet.*
4. *Use of Electronic/Manual Calculator is prohibited.*
5. ***No part of the Test Booklet and Answer Sheet shall be detached under any circumstances.***
6. *The candidates will write the Correct Test Booklet Code as given in the Test Booklet/Answer Sheet in the Attendance Sheet.*
7. *Compensatory time of one hour five minutes will be provided for the examination of three hours and 20 minutes duration, whether such candidate (having a physical limitation to write) uses the facility of scribe or not.*

PHYSICS

[redacted]

Q. 1. A football player is moving southward and suddenly turns eastward with the same speed to avoid an opponent. The force that acts on the player while turning is

(1) along north-ward **(2)** along north-east
(3) along south-west **(4)** along east-ward

Q. 2. An ac source is connected to a capacitor C. Due to decrease in its operating frequency

(1) displacement current increases.
(2) displacement current decreases.
(3) capacitive reactance remains constant
(4) capacitive reactance decreases.

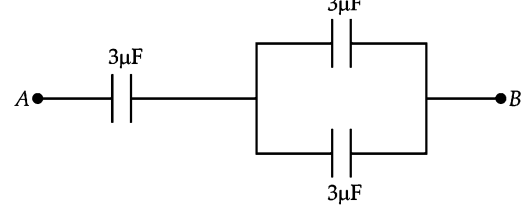
Q. 3. The half life of a radioactive substance is 20 minutes. In how much time, the activity of substance drops to $\left(\frac{1}{16}\right)^{th}$ of its initial value?

(1) 40 minutes **(2)** 60 minutes
(3) 80 minutes **(4)** 20 minutes

Q. 4. The net magnetic flux through any closed surface is

(1) Positive **(2)** Infinity
(3) Negative **(4)** Zero

Q. 5. The equivalent capacitance of the system shown in the following circuit is



(1) 3 μF **(2)** 6 μF
(3) 9 μF **(4)** 2 μF

Q. 6. The ratio of frequencies of fundamental harmonic produced by an open pipe to that of closed pipe having the same length is

(1) 2 :1 **(2)** 1 :3

(3) 3 :1 **(4)** 1 :2

Q. 7. In hydrogen spectrum, the shortest wavelength in the Balmer series is λ. The shortest wavelength in the Bracket series is

(1) 4 λ **(2)** 9 λ

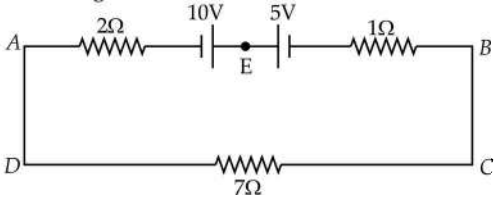
(3) 16 λ **(4)** 2 λ

Q. 8. A 12 V, 60 W lamp is connected to the secondary of a step down transformer, whose primary is connected to ac mains of 220 V. Assuming the transformer to be ideal, what is the current in the primary winding?

(1) 2.7 A **(2)** 3.7 A

(3) 0.37 A **(4)** 0.27 A

Q. 9. The magnitude and direction of the current in the following circuit is



(1) 0.5 A from *A* to *B* through *E*

(2) $\frac{5}{9}$ A from *A* to *B* through *E*

(3) 1.5 A from *B* to *A* through *E*

(4) 0.2 A from *B* to *A* through *E*

Q. 10. The magnetic energy stored in an inductor of inductance 4 μH carrying a current of 2 A is

(1) 4 mJ **(2)** 8 mJ

(3) 8 μJ **(4)** 4 μJ

Q. 11. The potential energy of a long spring when stretched by 2 cm is U. If the spring is stretched by 8 cm, potential energy stored in it will be

(1) 4U **(2)** 8U

(3) 16U **(4)** 2U

Q. 12. The errors in the measurement which arise due to unpredictable fluctuations in temperature and voltage supply are

(1) Personal errors

(2) Least count errors

(3) Random errors

(4) Instrumental errors

Q. 13. Resistance of a carbon resistor determined from colour codes is (22000 ± 5%)Ω. The colour of third band must be

(1) Green **(2)** Orange

(3) Yellow **(4)** Red

Q. 14. A metal wire has mass (0.4 ± 0.002) g, radius (0.3 ± 0.001) mm and length (5 ± 0.02) cm. The maximum possible percentage error in the measurement of density will nearly be

(1) 1.3% **(2)** 1.6%

(3) 1.4% **(4)** 1.2%

Q. 15. Light travels a distance x in time t_1 in air and $10x$ in time t_2 in another denser medium. What is the critical angle for this medium?

(1) $\sin^{-1}\left(\frac{10\,t_2}{t_1}\right)$ **(2)** $\sin^{-1}\left(\frac{t_1}{10\,t_2}\right)$

(3) $\sin^{-1}\left(\frac{10\,t_1}{t_2}\right)$ **(4)** $\sin^{-1}\left(\frac{t_2}{t_1}\right)$

Q. 16. In a plane electromagnetic wave travelling in free space, the electric field component oscillates sinusoidally at a frequency of 2.0×10^{10} Hz and amplitude 48 Vm^{-1}. Then the amplitude of oscillating magnetic field is (Speed of light in free space = 3×10^8 m s^{-1})

(1) 1.6×10^{-8}T **(2)** 1.6×10^{-7}T

(3) 1.6×10^{-6}T **(4)** 1.6×10^{-9}T

Q. 17. The ratio of radius of gyration of a solid sphere of mass *M* and radius *R* about its own axis to the radius of gyration of the thin hollow sphere of same mass and radius about its axis is

(1) 5 : 3 **(2)** 2 : 5

(3) 5 : 2 **(4)** 3 : 5

Q. 18. The amount of energy required to form a soap bubble of radius 2 cm from a soap solution is nearly (surface tension of soap solution = 0.03 N m^{-1})

(1) 5.06×10^{-4}J **(2)** 3.01×10^{-4}J

(3) 50.1×10^{-4}J **(4)** 30.16×10^{-4}J

Q. 19. In a series *LCR* circuit, the inductance *L* is 10 mH, capacitance *C* is 1 μF and resistance *R* is 100Ω. The frequency at which resonance occurs is

(1) 15.9 kHz **(2)** 1.59 rad/s

(3) 1.59 kHz **(4)** 15.9 rad/s

Q. 20. If $\oint_s \vec{E}.\vec{dS} = 0$ over a surface, then

(1) the magnitude of electric field on the surface is constant.

(2) all the charges must necessarily be inside the surface.

(3) the electric field inside the surface is necessarily uniform.

(4) the number of flux lines entering the surface must be equal to the number of flux lines leaving it.

Q. 21. The venturi-meter works on

(1) Bernoulli's principle

(2) The principle of parallel axes

(3) The principle of perpendicular axes

(4) Huygen's principle

Q. 22. Two bodies of mass m and $9m$ are placed at a distance *R*. The gravitational potential on the line joining the bodies where the gravitational field equals zero, will be (*G* = gravitational constant)

(1) $-\frac{12\,Gm}{R}$ **(2)** $-\frac{16\,Gm}{R}$

(3) $-\frac{20\,Gm}{R}$ **(4)** $-\frac{8\,Gm}{R}$

Q. 23. The temperature of a gas is –50° C. To what temperature the gas should be heated so that the rms speed is increased by 3 times?

(1) 3295° C **(2)** 3097 K

(3) 223 K **(4)** 669° C

Q. 24. The minimum wavelength of *X*-rays produced by an electron accelerated through a potential difference of *V* volts is proportional to

(1) $\frac{1}{V}$ **(2)** $\frac{1}{\sqrt{V}}$

(3) V^2 **(4)** $\sqrt{V}$

Q. 25. A vehicle travels half the distance with speed v and the remaining distance with speed 2 v. Its average speed is

(1) $\frac{2v}{3}$ **(2)** $\frac{4v}{3}$

(3) $\frac{3v}{4}$ **(4)** $\frac{v}{3}$

Q. 26. The angular acceleration of a body, moving along the circumference of a circle, is

(1) along the radius towards the centre

(2) along the tangent to its position

(3) along the axis of rotation

(4) along the radius, away from centre

Q. 27. Let a wire be suspended from the ceiling (rigid support) and stretched by a weight *W* attached at its free end. The longitudinal stress at any point of cross-sectional area *A* of the wire is

(1) $\frac{W}{A}$ **(2)** $\frac{W}{2A}$

(3) Zero **(4)** $\frac{2W}{A}$

Q. 28. A full wave rectifier circuit consists of two p-n junction diodes, a centre-tapped transformer, capacitor and a load resistance. Which of these components remove the ac ripple from the rectified output?

(1) p-n junction diodes

(2) Capacitor

(3) Load resistance

(4) A centre-tapped transformer

Q. 29. An electric dipole is placed at an angle of 30° with an electric field of intensity 2×10^5 NC^{-1}. It experiences a torque equal to 4 N m. Calculate the magnitude of charge on the dipole, if the dipole length is 2 cm.

(1) 6 mC **(2)** 4 mC

(3) 2 mC **(4)** 8 mC

Q. 30. A Carnot engine has an efficiency of 50% when its source is at a temperature 327° C. The temperature of the sink is

(1) 15° C **(2)** 100° C

(3) 200° C **(4)** 27° C

Q. 31. For Young's double slit experiment, two statements are given below:

Statement 1: If screen is moved away from the plane of slits, angular separation of the fringes remains constant.

Statement II: If the monochromatic source is replaced by another monochromatic source of larger wavelength, the angular separation of fringes decreases.

In the light of the above statements, choose the correct answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is true but **Statement II** is false.

(3) **Statement I** is false but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 32. The work functions of Caesium (Cs), Potassium (K) and Sodium (Na) are 2.14 eV, 2.30 eV and 2.75 eV respectively. If incident electromagnetic radiation has an incident energy of 2.20 eV, which of these photosensitive surfaces may emit photoelectrons?

(1) Both Na and K **(2)** K only

(3) Na only **(4)** Cs only

Q. 33. Given below are two statements:

Statement 1: Photovoltaic devices can convert optical radiation into electricity.

Statement II: Zener diode is designed to operate under reverse bias in breakdown region.

In the light of the above statements, choose the ***most appropriate*** answer from the options given below:

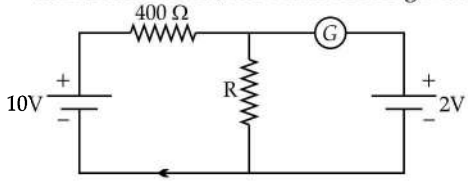
(1) Both **Statement I** and **Statement II** are incorrect.

(2) **Statement I** is correct but **Statement II** is incorrect.

(3) **Statement I** is incorrect but **Statement II** is correct.

(4) Both **Statement I** and **Statement II** are correct.

Q. 34. If the galvanometer *G* does not show any deflection in the circuit shown, the value of *R* is given by



(1) 50 Ω **(2)** 100 Ω

(3) 400 Ω **(4)** 200 Ω

Q. 35. A bullet is fired from a gun at the speed of 280 m s^{-1} in the direction 30° above the horizontal. The maximum height attained by the bullet is (g=9.8 m s^{-2}, sin 30° = 0.5)

(1) 2000 m **(2)** 1000 m

(3) 3000 m **(4)** 2800 m

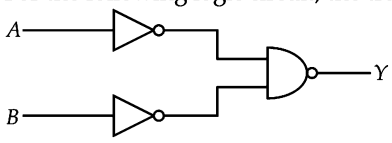
[redacted]

Q. 36. A satellite is orbiting just above the surface of the earth with period *T*. If *d* is the density of the earth and *G* is the universal constant of gravitation, the quantity $\frac{3\pi}{Gd}$ represents

(1) T^2 **(2)** T^3

(3) $\sqrt{T}$ **(4)** T

Q. 37. For the following logic circuit, the truth table is



(1)

A	B	Y
0	0	0
0	1	1
1	0	1
1	1	1

(2)

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	0

(3)

A	B	Y
0	0	0
0	1	0
1	0	0
1	1	1

(4)

A	B	Y
0	0	1
0	1	1
1	0	1
1	1	0

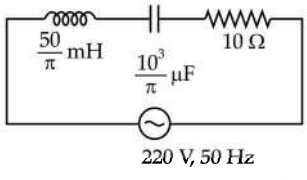
Q. 38. The radius of inner most orbit of hydrogen atom is 5.3×10^{-11} m. What is the radius of third allowed orbit of hydrogen atom?

(1) 1.06 Å **(2)** 1.59 Å
(3) 4.77 Å **(4)** 0.53 Å

Q. 39. A wire carrying a current I along the positive x-axis has length L. It is kept in a magnetic field $\vec{B} = \left(2\hat{i} + 3\hat{j} - 4\hat{k}\right)$ T. The magnitude of the magnetic force acting on the wire is

(1) $\sqrt{5}\,IL$ **(2)** $5\,IL$
(3) $\sqrt{3}\,IL$ **(4)** $3\,IL$

Q. 40. The net impedance of circuit (as shown in figure) will be

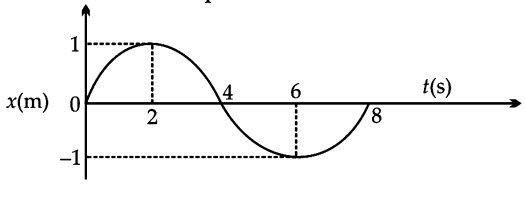


(1) 15 Ω **(2)** $5\sqrt{5}$ Ω
(3) 25 Ω **(4)** $10\sqrt{2}$ Ω

Q. 41. 10 resistors, each of resistance R are connected in series to a battery of emf E and negligible internal resistance. Then those are connected in parallel to the same battery, the current is increased n times. The value of n is

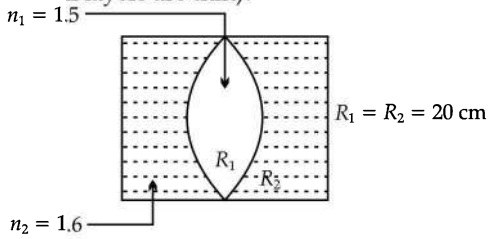
(1) 100 **(2)** 1
(3) 1000 **(4)** 10

Q. 42. The x-t graph of a particle performing simple harmonic motion is shown in the figure. The acceleration of the particle at $t = 2$ s is



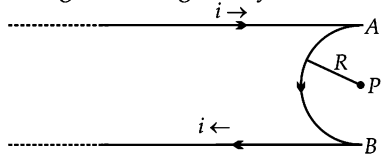
(1) $-\frac{\pi^2}{8}$ m s^{-2} **(2)** $\frac{\pi^2}{16}$ m s^{-2}
(3) $-\frac{\pi^2}{16}$ m s^{-2} **(4)** $\frac{\pi^2}{8}$ m s^{-2}

Q. 43. In the figure shown here, what is the equivalent focal length of the combination of lenses (Assume that all layers are thin)?



(1) − 40 cm **(2)** − 100 cm
(3) − 50 cm **(4)** 40 cm

Q. 44. A very long conducting wire is bent in a semi-circular shape from A to B as shown in figure. The magnetic field at point P for steady current configuration is given by



(1) $\frac{\mu_0 i}{4R}$ pointed away from the page

(2) $\frac{\mu_0 i}{4R}\left[1 - \frac{2}{\pi}\right]$ pointed away from page

(3) $\frac{\mu_0 i}{4R}\left[1 - \frac{2}{\pi}\right]$ pointed into the page

(4) $\frac{\mu_0 i}{4R}$ pointed into the page

Q. 45. Calculate the maximum acceleration of a moving car so that a body lying on the floor of the car remains stationary. The coefficient of static friction between the body and the floor is 0.15 (g = 10 ms^{-2})

(1) 150 ms^{-2} **(2)** 1.5 ms^{-2}
(3) 50 ms^{-2} **(4)** 1.2 ms^{-2}

Q. 46. Two thin lenses are of same focal lengths (f), but one is convex and the other one is concave. When they are placed in contact with each other, the equivalent focal length of the combination will be

(1) $\frac{f}{4}$ **(2)** $\frac{f}{2}$
(3) Infinite **(4)** Zero

Q. 47. A horizontal bridge is built across a river. A student standing on the bridge throws a small ball vertically upwards with a velocity 4 m s^{-1}. The ball strikes the water surface after 4 s. The height of bridge above water surface is (Take g = 10 ms^{-2})

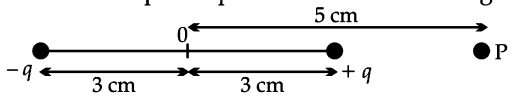
(1) 60 m **(2)** 64 m
(3) 68 m **(4)** 56 m

Q. 48. The resistance of platinum wire at 0°C is 2Ω and 6.8Ω at 80°C. The temperature coefficient of resistance of the wire is

(1) $3 \times 10^{-3}\ °C^{-1}$ (2) $3 \times 10^{-2}\ °C^{-1}$

(3) $3 \times 10^{-1}\ °C^{-1}$ (4) $3 \times 10^{-4}\ °C^{-1}$

Q. 49. An electric dipole is placed as shown in the figure.



The electric potential (in 10^2 V) at point P due to the dipole is (ε_0 = permittivity of free space and $\frac{1}{4\pi\varepsilon_0} = K$)

(1) $\left(\frac{5}{8}\right)$ qK (2) $\left(\frac{8}{5}\right)$ qK

(3) $\left(\frac{8}{3}\right)$ qK (4) $\left(\frac{3}{8}\right)$ qK

Q. 50. A bullet from a gun is fired on a rectangular wooden block with velocity u. When bullet travels 24 cm through the block along its length horizontally, velocity of bullet becomes $\frac{u}{3}$. Then it further penetrates into the block in the same direction before coming to rest exactly at the other end of the block. The total length of the block is

(1) 24 cm (2) 28 cm

(3) 30 cm (4) 27 cm

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CHEMISTRY

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Q. 51. The element expected to form largest ion to achieve the nearest noble gas configuration is

(1) F (2) N

(3) Na (4) O

Q. 52. In Lassaigne's extract of an organic compound, both nitrogen and sulphur are present, which gives blood red colour with Fe^{3+} due to the formation of

(1) NaSCN

(2) $[Fe(CN)_5\ NOS]^{4-}$

(3) $[Fe(SCN)]^{2+}$

(4) $Fe_4[Fe(CN)_6]_3 \cdot xH_2O$

Q. 53. The relation between n_m, (n_m = the number of permissible values of magnetic quantum number (m)) for a given value of azimuthal quantum number (l), is

(1) $l = 2n_m + 1$ (2) $n_m = 2l^2 + 1$

(4) $n_m = l + 2$ (4) $l = \frac{n_m - 1}{2}$

Q. 54. Which one is an example of heterogenous catalysis?

(1) Hydrolysis of sugar catalysed by H^+ ions.

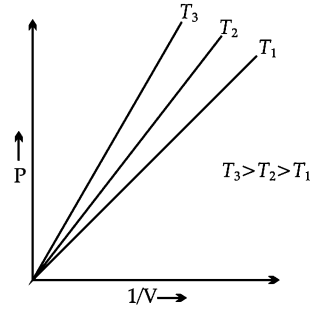
(2) Decomposition of ozone in presence of nitrogen monoxide.

(3) Combination between dinitrogen and dihydrogen to form ammonia in the presence of finely divided iron.

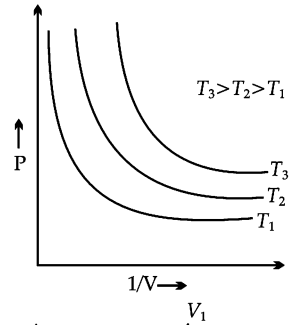
(4) Oxidation of sulphur dioxide into sulphur trioxide in the presence of oxides of nitrogen.

Q. 55. Which amongst the following options is **correct** graphical representation of Boyle's Law?

(1)



(2)



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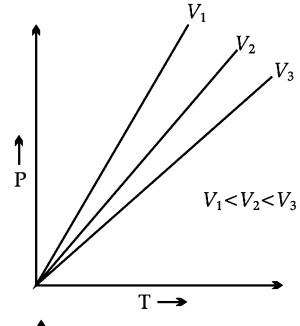

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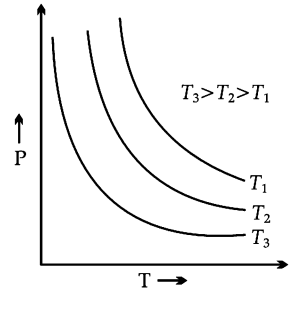
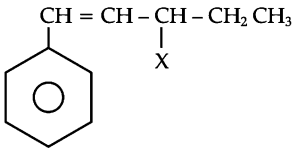


Q. 56. The given compound

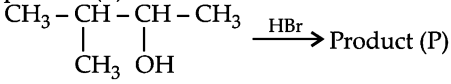


is an example of _____.

(1) aryl halide **(2)** allylic halide

(3) vinylic halide **(4)** benzylic halide

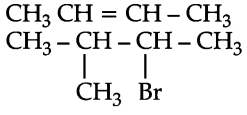
Q. 57. Consider the following reaction and identify the product (P).



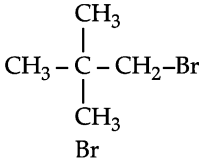
3 – Methylbutan –2 – ol

(1) $CH_3CH = CH - CH_3$

(2)



(3)



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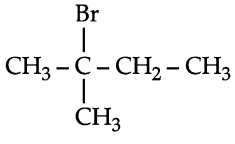


Q. 58. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R:**
Assertion A: Helium is used to dilute oxygen in diving apparatus.
Reasons R: Helium has high solubility in O_2.
In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 59. The conductivity of centimolar solution of KCl at 25°C is 0.0210 ohm^{-1} cm^{-1} and the resistance of the cell containing the solution at 25°C is 60 ohm. The value of cell constant is

(1) 3.28 cm^{-1} **(2)** 1.26 cm^{-1}

(3) 3.34 cm^{-1} **(4)** 1.34 cm^{-1}

Q. 60. The number of σ bonds, π bonds and lone pair of electrons in pyridine, respectively are

(1) 12, 3, 0 **(2)** 11, 3, 1

(3) 12, 2, 1 **(4)** 11, 2, 0

Q. 61. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R:**
Assertion A: Metallic sodium dissolves in liquid ammonia giving a deep blue solution, which is paramagnetic.
Reasons R: The deep blue solution is due to the formation of amide.
In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 62. The **right** option for the mass of CO_2 produced by heating 20 g of 20% pure limestone is (Atomic mass of Ca = 40)

$$\left[CaCO_3 \xrightarrow{1200\ K} CaO + CO_2\right]$$

(1) 1.76 g **(2)** 2.64 g

(3) 1.32 g **(4)** 1.12 g

Q. 63. Intermolecular forces are forces of attraction and repulsion between interacting particles that will include

A. dipole-dipole forces.

B. dipole-induced dipole forces.

C. hydrogen bonding.

D. covalent bonding.

E. dispersion forces.

Choose the **most appropriate** answer from the options given below :

(1) A, B, C, D are correct.

(2) A, B, C, E are correct.

(3) A, C, D, E are correct.

(4) B, C, D, E are correct.

Q. 64. For a certain reaction, the rate = $k[A]^2[B]$, when the initial concentration of A is tripled keeping concentration of B constant, the initial rate would

(1) increase by a factor of six.

(2) increase by a factor of nine.

(3) increase by a factor of three.

(4) decrease by a factor of nine.

Q. 65. Taking stability as the factor, which one of the following represents **correct** relationship?

(1) $InI_3 > InI$

(2) $AlCl > AlCl_3$

(3) $TlI > TlI_3$

(4) $TlCl_3 > TlC$

Q. 66. Which of the following statements are NOT correct?

A. Hydrogen is used to reduce heavy metal oxides to metals.

B. Heavy water is used to study reaction mechanism.

C. Hydrogen is used to make saturated fats from oils.

D. The H-H bond dissociation enthalpy is lowest as compared to a single bond between two atoms of any element.

E. Hydrogen reduces oxides of metals that are more active than iron.

Choose the **most appropriate** answer from the options given below:

(1) B, D only

(2) D, E only

(3) A, B, C only

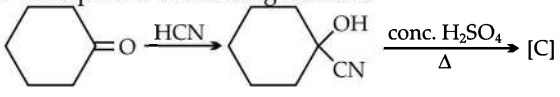
(4) B, C, D, E only

Q. 67. Weight (g) of two moles of the organic compound, which is obtained by heating sodium ethanoate with sodium hydroxide in presence of calcium oxide is

(1) 32 **(2)** 30

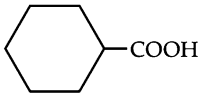
(3) 18 **(4)** 16

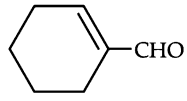
Q. 68. Complete the following reaction:

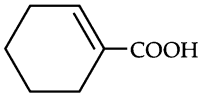


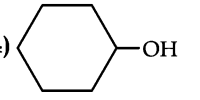
[A] [B]

[C] is _____.

(1) 

(2) 

(3) 

(4) 

Q. 69. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R:**

Assertion A: In equation $\Delta_r G = -nFE_{cell}$, value of $\Delta_r G$ depends on n.

Reason R: E_{cell} is an intensive property and $\Delta_r G$ is an extensive property.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 70. Amongst the following, the total number of species NOT having eight electrons around central atom in its outer most shell, is

NH_3, $AlCl_3$, $BeCl_2$, CCl_4, PCl_5 :

(1) 2 **(2)** 4

(3) I **(4)** 3

Q. 71. The **correct** order of energies of molecular orbitals of N_2 molecule, is:

(1) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \sigma 2p_z < (\pi 2p_x = \pi 2p_y) < (\pi^* 2p_x = \pi^* 2p_y) < \sigma^* 2p_z$

(2) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \sigma 2p_z < \sigma^* 2p_z < (\pi 2p_x = \pi 2p_y) < (\pi^* 2p_x = \pi^* 2p_y)$

(3) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < (\pi 2p_x = \pi 2p_y) < (\pi^* 2p_x = \pi^* 2p_y) < \sigma 2p_z < \sigma^* 2p_z$

(4) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < (\pi 2p_x = \pi 2p_y) < \sigma 2p_z < (\pi^* 2p_x = \pi^* 2p_y) < \sigma^* 2p_z$

Q. 72. A compound is formed by two elements A and B. The element B forms cubic close packed structure and atoms of A occupy 1/3 of tetrahedral voids. If the formula of the compound is A_xB_y, then the value of x + y is in option

(1) 4 **(2)** 3

(3) 2 **(4)** 5

Q. 73. The stability of Cu^{2+} is more than Cu^+ salts in aqueous solution due to

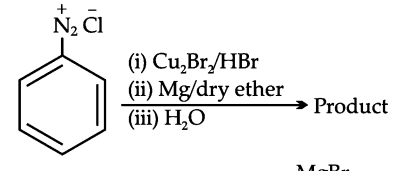
(1) enthalpy of atomization.

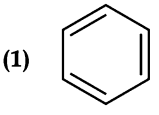
(2) hydration energy.

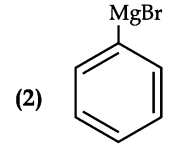
(3) second ionisation enthalpy.

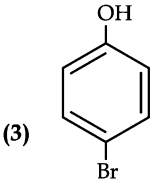
(4) first ionisation enthalpy.

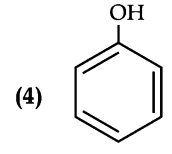
Q. 74. Identify the product in the following reaction



(1) 

(2) 

(3) 

(4) 

Q. 75. Given below are two statements:

Statement. I: A unit formed by the attachment of a base to 1′ position of sugar is known as nucleoside

Statement II: When nucleoside is linked to phosphorous acid at 5′-position of sugar moiety, we get nucleotide.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both Statement **I** and Statement **II** are false.

(2) Statement **I** is true but Statement **II** is false.

(3) Statement **I** is false but Statement **II** is true.

(4) Both Statement **I** and Statement **II** are true.

Q. 76. Which one of the following statements is **correct**?

(1) All enzymes that utilise ATP in phosphate transfer require Ca as the cofactor.

(2) The bone in human body is an inert and unchanging substance.

(3) Mg plays roles in neuromuscular function and interneuronal transmission.

(4) The daily requirement of Mg and Ca in the human body is estimated to be 0.2 - 0.3 g.

Q. 77. Which of the following reactions will NOT give primary amine as the product?

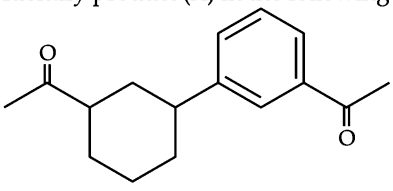
(1) $CH_3CN \xrightarrow[(ii)\ H_3O^{\oplus}]{(i)\ LiAlH_4}$ Product

(2) $CH_3NC \xrightarrow[(ii)\ H_3O^{\oplus}]{(i)\ LiAlH_4}$ Product

(3) $CH_3CONH_2 \xrightarrow[(ii)\ H_3O^{\oplus}]{(i)\ LiAlH_4}$ Product

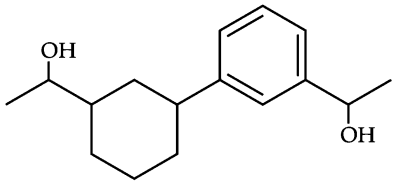
(4) $CH_3\,CONH_2 \xrightarrow{Br_2\ /\ KOH}$ Product

Q. 78. Identify product (A) in the following reaction:

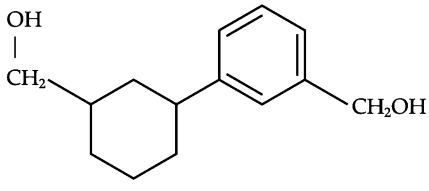


$\xrightarrow[Conc.HCl]{Zn-Hg}$ (A) + $2H_2O$

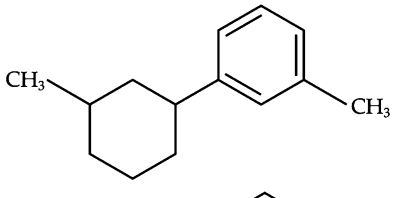
(1)



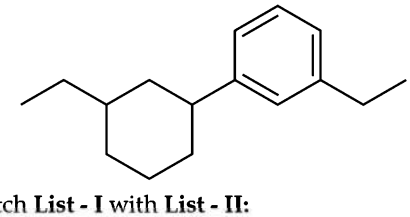
(2)



(3)



(4)



Q. 79. Match **List - I** with **List - II:**

List - I	List-II
A. Coke	I. Carbon atoms are sp^3 hybridised.
B. Diamond	II. Used as a dry lubricant
C. Fullerene	III. Used as a reducing agent
D. Graphite	IV. Cage like molecules

Choose the **correct** answer from the options given below:

(1) A-IV, B-I, C-II, D-III

(2) A-III, B-I, C-IV, D-II

(3) A-III, B-IV, C-I, D-II

(4) A-II, B-IV, C-I, D-III

Q. 80. Amongst the given options which of the following molecules/ion acts as a Lewis acid?

(1) H_2O **(2)** BF_3

(3) OH^- **(4)** NH_3

Q. 81. Which amongst the following molecules on polymerization produces neoprene?

(1) $H_2C = \overset{Cl}{\overset{|}{C}} - CH = CH_2$

(2) $H_2C = \overset{CH_3}{\overset{|}{C}} - C \equiv CH$

(3) $H_2C = \overset{CH_3}{\overset{|}{C}} - CH = CH_2$

(4) $H_2C = CH - CH = CH_2$

Q. 82. Some tranquilizers are listed below. Which one from the following belongs to barbiturates?

(1) Meprobamate

(2) Valium

(3) Veronal

(4) Chlordiazepoxide

Q. 83. Homoleptic complex from the following complexes is

(1) Diamminechloridonitrito - N- platinum (II)

(2) Pentaamminecarbonatocobalt (III) chloride

(3) Triamminetriaquachromium (III) chloride

(4) Potassiumtrioxalatoaluminate (III)

Q. 84. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R:**

Assertion A: A reaction can have zero activation energy.

Reasons R: The minimum extra amount of energy absorbed by reactant molecules so that their energy becomes equal to threshold value, is called activation energy.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 85. Select the **correct** statements from the following:

A. Atoms of all elements are composed of two fundamental particles.

B. The mass of the electron is 9.10939×10^{-31} kg.

C. All the isotopes of a given element show same chemical properties.

D. Protons and electrons are collectively known as nucleons.

E. Dalton's atomic theory, regarded the atom as an ultimate particle of matter.

Choose the **correct** answer from the options given below:

(1) C, D and E only

(2) A and E only

(3) B, C and E only

(4) A, B and C only

[redacted]

Q. 86. Match **List-I** with **List - II:**

List-I (Oxoacids of Sulphur)		List-II (Bonds)	
A.	*Peroxodisulphuric acid*	I.	Two S-OH, Four S=O, One S-O-S

B.	Sulphuric acid	**II.**	Two S-OH, One S=O
C.	Pyrosulphuric acid	**III.**	Two S-OH, Four S=O, One S-O-O-S
D.	Sulphurous acid	**IV.**	Two S-OH, Two S=O

Choose the **correct** answer from the options given below:

(1) A-III, B-IV, C-I, D-II

(2) A-I, B-III, C-IV, D-II

(3) A-III, B-IV, C-II, D-I

(4) A-1, B-III, C-II, D-IV

Q. 87. On balancing the given redox reaction,

$$a\,Cr_2O_7^{2-} + b\,SO_3^{2-}\,(aq) + c\,H^+\,(aq) \rightarrow 2a\,Cr^{3+}\,(aq) + b\,SO_4^{2-}\,(aq) + \frac{c}{2}\,H_2O\,(l)$$

the coefficients a, b and c are found to be, respectively

(1) 3, 8, 1 **(2)** 1, 8, 3

(3) 8, 1, 3 **(4)** 1, 3, 8

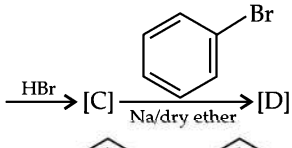
Q. 88. What fraction of one edge centred octahedral void lies in one unit cell of fcc?

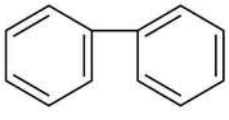
(1) $\frac{1}{3}$ **(2)** $\frac{1}{4}$

(3) $\frac{1}{12}$ **(4)** $\frac{1}{2}$

Q. 89. Identify the final product [D] obtained in the following sequence of reactions.

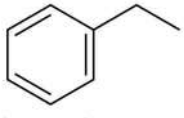
$$CH_3CHO \xrightarrow[ii)\ H_3O^+]{i)\ LiAlH_4} [A] \xrightarrow[\Delta]{H_2SO_4} [B]$$



(1) 

(2) C_4H_{10}

(3) $HC \equiv C^{\ominus}\,Na^+$

(4) 

Q. 90. Which complex compound is most stable?

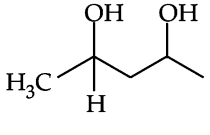
(1) $[Co(NH_3)_3(NO_3)_3]$

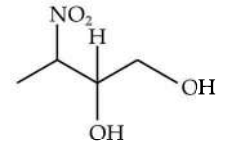
(2) $[CoCl_2\,(en)_2]\,NO_3$

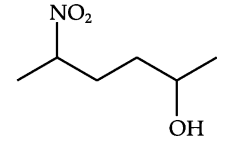
(3) $[Co(NH_3)_6]_2(SO_4)_3$

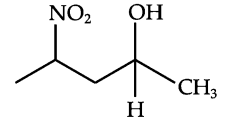
(4) $[Co(NH_3)_4\,(H_2O)Br](NO_3)_2$

Q. 91. Which amongst the following will be most readily dehydrated under acidic conditions ?

(1) 

(2) 

(3) 

(4) 

Q. 92. Given below are two statements:

Statement I: The nutrient deficient water bodies lead to eutrophication.

Statement II: Eutrophication leads to decrease in the level of oxygen in the water bodies.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is correct but **Statement II** is false.

(3) **Statement I** is incorrect but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 93. The reaction that does **NOT** take place in a blast furnace between 900 K to 1500 K temperature range during extraction of iron is

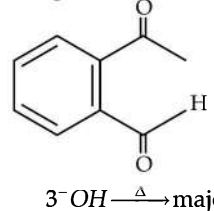
(1) $FeO + CO \rightarrow Fe + CO_2$

(2) $C + CO_2 \rightarrow 2CO$

(3) $CaO + SiO_2 \rightarrow CaSiO_3$

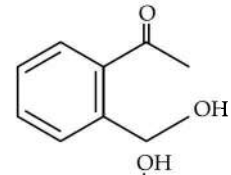
(4) $Fe_2O_3 + CO \rightarrow 2FeO + CO_2$

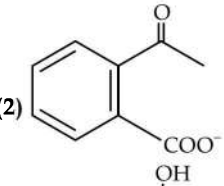
Q. 94. Identify the major product obtained in the following reaction :

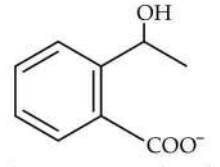


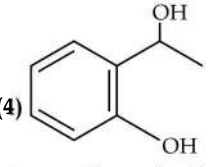
$+ 2[Ag(NH_3)_2]^+$

$3^-OH \xrightarrow{\Delta}$ major product

(1) 

(2) 

(3) 

(4) 

Q. 95. Which amongst the following options is the **correct** relation between change in enthalpy and change in internal energy?

(1) $\Delta H = \Delta U + \Delta n_g RT$

(2) $\Delta H - \Delta U = -\Delta nRT$

(3) $\Delta H + \Delta U = \Delta nR$

(4) $\Delta H = \Delta U - \Delta n_g RT$

Q. 96. Which of the following statements are **INCORRECT**?

A. All the transition metals except scandium form MO oxides which are ionic.

B. The highest oxidation number corresponding to the group number in transition metal oxides is attained in Sc_2O_3 to Mn_2O_7.

C. Basic character increases from V_2O_3 to V_2O_4 to V_2O_5.

D. V_2O_4 dissolves in acids to give VO_4^{3-} salts.

E. CrO is basic but Cr_2O_3 is amphoteric.

Choose the **correct** answer from the options given below:

(1) B and D only

(2) C and D only

(3) B and C only

(4) A and E only

Q. 97. The equilibrium concentrations of the species in the reaction $A + B \rightleftharpoons C + D$ are 2, 3, 10 and 6 mol L^{-1}, respectively at 300 K. $\Delta G°$ for the reaction is (R = 2 cal/mol K)

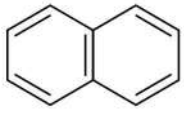
(1) – 137.26 cal

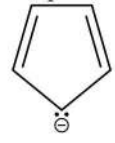
(2) – 1381.80 cal

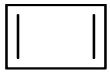
(3) – 13.73 cal

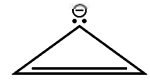
(4) 1372.60 cal

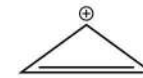
Q. 98. Consider the following compounds/species:

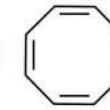
(i) 

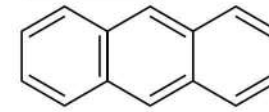
(ii) 

(iii) 

(iv) 

(v) 

(vi) 

(vi) 

The number of compounds/species which obey Huckel's rule is _____.

(1) 6 **(2)** 2

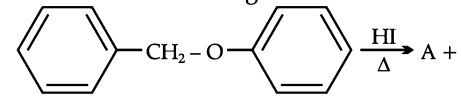
(3) 5 **(4)** 4

Q. 99. Pumice stone is an example of

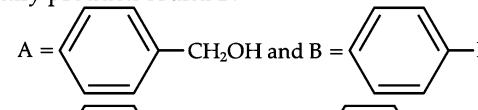
(1) gel **(2)** solid sol

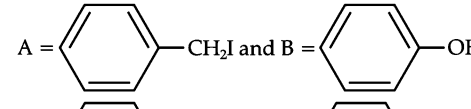
(3) foam **(4)** sol

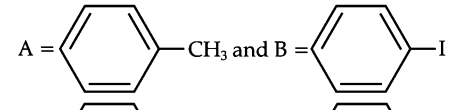
Q. 100. Consider the following reaction:

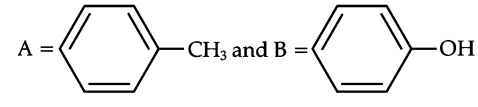


Identify products A and B.

(1) 

(2) 

(3) 

(4) 

BOTANY

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Q. 101. The phenomenon of pleiotropism refers to

(1) presence of two alleles, each of the two genes controlling a single trait.

(2) a single gene affecting multiple phenotypic expression.

(3) more than two genes affecting a single character.

(4) presence of several alleles of a single gene controlling a single crossover.

Q. 102. In tissue culture experiments, leaf mesophyll cells are put in a culture medium to form callus. This phenomenon may be called as :

(1) Dedifferentiation

(2) Development

(3) Senescence

(4) Differentiation

Q. 103. Movement and accumulation of ions across a membrane against their concentration gradient can be explained by

(1) Facilitated Diffusion

(2) Passive Transport

(3) Active Transport

(4) Osmosis.

Q. 104. Among The Evil Quartet', which one is considered the most important cause driving extinction of species?

(1) Over exploitation for economic gain

(2) Alien species invasions

(3) Co-extinctions

(4) Habitat loss and fragmentation

Q. 105. Upon exposure to UV radiation, DNA stained with ethidium bromide will show

(1) Bright blue colour

(2) Bright yellow colour

(3) Bright orange colour

(4) Bright red colour

Q. 106. Given below are two statements: One is labelled as **Assertion A** and the other is labelled as **Reason R:**
Assertion A: The first stage of gametophyte in the life cycle of moss is protonema stage.
Reason R: Protonema develops directly from spores produced in capsule.
In the light of the above statements, choose the **most appropriate** answer from the options given below:

(1) Both **A** and **R** are correct but **R** is NOT the correct explanation of **A**.

(2) **A** is correct but **R** is not correct.

(3) **A** is not correct but **R** is correct.

(4) Both **A** and **R** are correct and **R** is the correct explanation of **A**.

Q. 107. What is the role of RNA polymerase III in the process of transcription in eukaryotes?

(1) Transcription of tRNA, 5 srRNA and snRNA

(2) Transcription of precursor of mRNA

(3) Transcription of only snRNAs

(4) Transcription of rRNAs (28S, 18S, and 5.8S)

Q. 108. The historic convention on Biological Diversity, 'The Earth Summit' was held in Rio de Janeiro in the year :

(1) 1992 **(2)** 1986

(3) 2002 **(4)** 1985

Q. 109. Given below are two statements: One is labelled as **Assertion A** and the other is labelled as **Reason R** :
Assertion A: ATP is used at two steps in glycolysis,
Reason R: First ATP is used in converting glucose into glucose-6-phosphate and second ATP is used in conversion of fructose-6- phosphate into fructose-1-6-diphosphate.
In the light of the above statements, choose the correct answer from the options given below:

(1) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 110. The thickness of ozone in a column of air in the atmosphere is measured in terms of:

(1) Decibels **(2)** Decameter

(3) Kilobase **(4)** Dobson units

Q. 111. Given below are two statements:
Statement I: Endarch and exarch are the terms often used for describing the position of secondary xylem in the plant body.
Statement II: Exarch condition is the most common feature of the root system.
In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is correct but **Statement II** false.

(3) **Statement I** is incorrect but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 112. Identify the pair of heterosporous pteridophytes among the following:

(1) *Selaginella* and *Salvinia*

(2) *Psilotum* and *Salvinia*

(3) *Equisetum* and *Salvinia*

(4) *Lycopodium* and *Selaginella*

Q. 113. Identify the **correct** statements :

A. Detrivores perform fragmentation.

B. The humus is further degraded by some microbes during mineralization.

C. Water soluble inorganic nutrients go down into the soil and get precipitated by a process called leaching.

D. The detritus food chain begins with living organisms.

E. Earthworms break down detritus into smaller particles by a process called catabolism.

Choose the **correct** answer from the options given below:

(1) B, C, D only **(2)** C, D, E only

(3) D, E, A only **(4)** A, B, C only

Q. 114. Axile placentation is observed in

(1) China rose, Beans and Lupin

(2) Tomato, *Dianthus* and Pea

(3) China rose, *Petunia* and Lemon

(4) Mustard, Cucumber and Primrose

Q. 115. Family Fabaceae differs from Solanaceae and Liliaceae. With respect to the stamens, pick out the characteristics specific to family Fabaceae but not found in Solanaceae or Liliaceae.

(1) Polyadelphous and epipetalous stamen

(2) Monoadelphous and Monothecou anthers

(3) Epiphyllous and Dithecous anthers

(4) Diadelphous and Dithecous anthers

Q. 116. The reaction centre in PS II has an absorption maxima at

(1) 700 nm **(2)** 660 nm

(3) 780 nm **(4)** 680 nm

Q. 117. During the purification process for recombinant DNA technology, addition of chilled ethanol precipitates out

(1) DNA **(2)** Histones

(3) Polysaccharides **(4)** RNA

Q. 118. Among eukaryotes, replication of DNA takes place in

(1) S phase **(2)** G_1 phase

(3) G_1 phase **(4)** M phase

Q. 119. The process of appearance of recombination nodules occurs at which sub stage of prophase I in meiosis?

(1) Pachytene **(2)** Diplotene

(3) Diakinesis **(4)** Zygotene

Q. 120. How many ATP and $NADPH_2$ are required for the synthesis of one molecule of glucose during calvin cycle?

(1) 18 ATP and 12 $NADPH_2$

(2) 12 ATP and 16 $NADPH_2$

(3) 18 ATP and 16 $NADPH_2$

(4) 12 ATP and 12 $NADPH_2$

Q. 121. In the equation

$$GPP - R = NPP$$

GPP is Gross Primary Productivity
NPP is Net Primary Productivity
R here is _____.

(1) Respiratory quotient

(2) Respiratory loss

(3) Reproductive allocation

(4) Photosynthetically active radiation

Q. 122. Large, colourful, fragrant flowers with nectar are seen in:

(1) bird pollinated plants

(2) bat pollinated plants

(3) wind pollinated plants

(4) insect pollinated plants

Q. 123. Spraying of which of the following phytohormone on juvenile conifers helps in hastening the maturity period, that leads to early seed production?

(1) Gibberellic Acid

(2) Zeatin

(3) Abscisic Acid

(4) Indole-3-butyric Acid

Q. 124. Which micronutrient is required for splitting of water molecule during photosynthesis?

(1) molybdenum **(2)** magnesium

(3) copper **(4)** manganese

Q. 125. Expressed Sequence Tags (ESTs) refers to

(1) All genes that are expressed as proteins.

(2) All genes whether expressed or unexpressed.

(3) Certain important expressed genes.

(4) All genes that are expressed as RNA.

Q. 126. Frequency of recombination between gene pairs on same chromosome as a measure of the distance between genes to map their position on chromosome, was used for the first time by

(1) Sutton and Boveri

(2) Alfred Sturtevant

(3) Henking

(4) Thomas Hunt Morgan

Q. 127. Given below are two statements:

Statement I: The forces generated by transpiration can lift a xylem-sized column of water over 130 meters height.

Statement II: Transpiration cools leaf surfaces sometimes 10 to 15 degrees, by evaporative cooling.

In the light of the above statements, choose the **most appropriate** answer from the options given below:

(1) Both **Statement I** and **Statement II** are incorrect.

(2) **Statement I** is correct but **Statement II** is incorrect.

(3) **Statement I** is incorrect but **Statement II** is correct.

(4) Both **Statement I** and **Statement II** are correct.

Q. 128. In gene gun method used to introduce alien DNA into host cells, microparticles of _____ metal are used.

(1) Zinc

(2) Tungsten or gold

(3) Silver

(4) Copper

Q. 129. Given below are two statements: One is labelled as **Assertion A** and the other is labelled as **Reason R**:

Assertion A: Late wood has fewer xylary elements with narrow vessels.

Reason R: Cambium is less active in winters.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 130. What is the function of tassels in the corn cob?

(1) To trap pollen grains

(2) To disperse pollen grains

(3) To protect seeds

(4) To attract insects

Q. 131. In angiosperm, the haploid, diploid and triploid structures of a fertilized embryo sac sequentially are:

(1) Antipodals, synergids, and primary endosperm nucleus

(2) Synergids, zygote and primary endosperm nucleus

(3) Synergids, antipodals and polar nuclei

(4) Synergids, primary endosperm nucleus and zygote

Q. 132. Which hormone promotes internode/ptiole elongation in deep water rice?

(1) Kinetin **(2)** Ethylene

(3) 2, 4-D **(4)** GA_3

Q. 133. Which of the following stages of meiosis involves division of centromere?

(1) Metaphase II **(2)** Anaphase II

(3) Telophase **(4)** Metaphase I

Q. 134. Unequivocal proof that DNA is the genetic material was first proposed by

(1) Alfred Hershey and Martha Chase

(2) Avery, Macleoid and McCarthy

(3) Wilkins and Franklin

(4) Frederick Griffith

Q. 135. Cellulose does not form blue colour with Iodine because

(1) It is a helical molecule.

(2) It does not contain complex helices and hence cannot hold iodine molecules.

(3) It breakes down when iodine reacts with it.

(4) It is a disaccharide.

Q. 136. Match **List I** with **List II:**

	List I		List II
A.	Oxidative decarboxylation	I.	Citrate synthase
B.	Glycolysis	II.	Pyruvate dehydrogenase
C.	Oxidative phosphorylation	III.	Electron transport system
D.	Tricarboxylic acid cycle	IV.	EMP pathway

Choose the correct answer from the option given below:

(1) A-II, B-IV, C-I, D-III
(2) A-III, B-I, C-II, D-IV
(3) A-II, B-IV, C-III, D-I
(4) A-III, B-IV, C-II, D-I

Q. 137. Which of the following combinations required for chemiosmosis?

(1) Membrane, proton pump, proton gradient, NADP synthase
(2) Proton pump, electron gradient, ATP synthase
(3) Proton pump, electron gradient, NADP synthase
(4) Membrane, proton pump, proton gradient, ATP synthase

Q.138. Melonate inhibits the growth of pathogenic bacteria by inhibiting the activity of

(1) Amylase
(2) Lipase
(3) Dinitrogenase
(4) Succinic dehydrogenase

Q. 139. Given below are two statements : One labelled as **Assertion A** and the other labelled as **Reason R:**
Assertion A: In gymnosperms the pollen grains are released from the microsporangium and carried by air currents.
Reason R: Air currents carry the pollen grains to the mouth of the archegonia where the male gametes are discharged and pollen tube is not formed.
In the light of the above statements, choose the **correct** answer from the options give below:

(1) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.
(2) **A** is true but **R** is false.
(3) **A** is false but **R** is true.
(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 140. Main steps in the formation of Recombinant DNA are given below. Arrange these steps in a correct sequence.

A. Insertion of recombinant DNA into the host cell.
B. Cutting of DNA at specific location by restriction enzyme.
C. Isolation of desired DNA fragment.
D. Amplification of gene of interest using PCR.

Choose the correct answer from the options given below:

(1) C, A, B, D
(2) C, B, D, A
(3) B, D, A, C
(4) B, C, D, A

Q. 141. Match **List I** with **List II** :

	List I		List II
A.	Cohesion	I.	More attraction in liquid phase
B.	Adhesion	II.	Mutual attraction among water molecules
C.	Surface tension	III.	Water loss in liquid phase
D.	Guttation	IV.	Attraction towards polar surfaces

Choose the **correct** answer from the options given below:

(1) A-IV, B-III, C-II, D-I
(2) A-III, B-I, C-IV, D-II
(3) A-II, B-I, C-IV, D-III
(4) A-II, B-IV, C-I, D-III

Q. 142. Which one of the following statements is **NOT** correct?

(1) Algal blooms caused by excess of organic matter in water improve water quality and promote fisheries.
(2) Water hyacinth grows abundantly in eutrophic water bodies and leads to an imbalance in the ecosystem dynamics of the water body.
(3) The amount of some toxic substances of industrial waste water increases in the organisms at successive trophic levels.
(4) The micro-organisms involved in biodegradation of organic matter in a sewage polluted water body consume a lot of oxygen causing the death of aquatic organisms.

Q. 143. How many different proteins does the ribosome consist of?

(1) 60
(2) 40
(3) 20
(4) 80

Q. 144. Match **List I** with **List II:**

	List I (Interaction)		List II (Species A and B)
A.	Mutualism	I.	+ (A), O(B)
B.	Commensalism	II.	- (A), O(B)
C.	Amensalism	III.	+ (A), −(B)
D.	Parasitism	IV.	+ (A), +(B)

Choose the **correct** answer from the options given below:

(1) A-IV, B-I, C-II, D-III
(2) A-IV, B-III, C-I. D-II
(3) A-III, B-I, C-IV, D-II
(4) A-IV, B-II, C-I, D-III

Q. 145. Given below are two statements : One is labelled as **Assertion A** and the other is labeled as **Reason R:**

Assertion A: A flower is defined as modified shoot wherein the shoot apical meristem changes to floral meristem.

Reason R: Internode of the shoot gets condensed to produce different floral appendages laterally at successive nodes instead of leaves.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 146. Identify the **correct** statements:

A. Lenticels are the lens-shaped openings permitting the exchange of gases.

B. Bark formed early in the season is called hard bark.

C. Bark is a technical term that refers to all tissues exterior to vascular cambium.

D. Bark refers to periderm and secondary phloem.

E. Phellogen is single-layered in thickness.

Choose the **correct** answer from the options given below:

(1) A and D only

(2) A, B and D only

(3) B and C only

(4) B, C and E only

Q. 147. Given below are two statements :

Statement I: Gause's 'Competitive Exclusion Principle states that two closely related species competing for the same resources cannot co-exist indefinitely and competitively inferior one will be eliminated eventually.

Statement II: In general, carnivores are more adversely affected by competition than herbivores.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is correct but **Statement II** is false.

(3) **Statement I** is incorrect but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 148. Which of the following statements are correct about Klinefelter's Syndrome?

A. This disorder was first described by Langdon Down (1866).

B. Such an individual has overall masculine development. However, the feminine development is also expressed.

C. The affected individual is short statured.

D. Physical, psychomotor and mental development is retarded.

E. Such individuals are sterile.

Choose the **correct** answer from the options given below:

(1) C and D only **(2)** B and E only

(3) A and E only **(4)** A and B only

Q. 149. Match **List I** with **List II:**

	List I		List II
A.	M Phase	I.	Proteins are synthesized
B.	G_2 Phase	II.	Inactive phase
C.	Quiescent stage	III.	Interval between mitosis and initiation of DNA replication
D.	G_1 Phase	IV.	Equational division

Choose the correct answer from the options given below:

(1) A-IV, B-II, C-I, D-III

(2) A-IV, B-I, C-II, D-III

(3) A-II, B-IV, C-I, D-III

(4) A-III, B-II, C-IV, D-I

Q. 150. Match List I with **List II:**

List I		List II	
A.	Iron	I.	Synthesis of auxin
B.	Zinc	II.	Component of nitrate reductase
C.	Boron	III.	Activator of catalase
D.	Molybdenum	IV.	Cell elongation and differentiation

Choose the correct answer from the options given below:

(1) A-II, B-III, C-IV, D-I

(2) A-III, B-I, C-IV, D-II

(3) A-II, B-IV, C-I, D-III

(4) A-III, B-II, C-I, D-IV

ZOOLOGY

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Q. 151. Given below are two statements:

Statement 1: In prokaryotes, the positively charged DNA is held with some negatively charged proteins in a region called nucleoid.

Statement II: In eukaryotes, the negatively charged DNA is wrapped around the positively charged histone octamer to form nucleosome.

In the light of the above statements, the **correct** answer from the options given below:

(1) Both the **statements I** and **Statement II** are false.

(2) **Statement I is** correct but **Statement II** is false.

(3) **Statement I** incorrect but **Statement II is true.**

(4) Both **Statement I** and **Statement II** are true.

Q. 152. In which blood corpuscles, the HIV undergoes replication and produces progeny viruses?

(1) B-lymphocytes (2) Basophils

(3) Eosinophils (4) T_H cells

Q. 153. Which of the following statements is correct?

(1) Biomagnification refers to increase in concentration of the toxicant at successive trophic levels.

(2) Presence of large amount of nutrients in water restricts 'Algal Bloom'.

(3) Algal Bloom decreases fish mortality.

(4) Eutrophication refers to increase in domestic sewage and waste water in lakes.

Q. 154. Match **List I** with **List II.**

	List I		List II
A.	P-wave	I.	Beginning of systole
B.	Q-wave	II.	Repolarisation of ventricles
X.	QRS complex	III.	Depolarisation of atria
Δ.	T-wave	iv.	Depolarisation of ventricles

Choose the **correct** answer from the options given below:

(1) A-IV, B-III, C-II, D-1

(2) A-II, B-IV, C-I, D-III

(3) A-1, B-II, C-III, D-IV

(4) A-III, B-I, C-IV, D-II

Q. 155. Broad palm with single palm crease is visible in a person suffering from-

(1) Turner's syndrome

(2) Klinefelter's syndrome

(3) Thalassemia

(4) Down's syndrome

Q. 156. Given below are two statements:

Statement I: RNA mutates at a faster rate.

Statement II: Viruses having RNA genome and shorter life span mutate and evolve faster.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is true but **Statement** is false.

(3) **Statement I** false but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 157. Given below are statements:

Statement I: Ligaments are dense irregular tissue.

Statement II: Cartilage is dense regular tissue.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is true but **Statement** is false.

(3) **Statement I** false but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 158. Given below are statements: one is labelled as **Assertion A** and the other is labelled as **Reason R.**

Assertion A: Nephrons are of two types: Cortical & Juxta medullary, based on their relative position in cortex and medulla.

Reason R: Juxta medullary nephrons have short loop of Henle whereas, cortical nephrons have longer loop of Henle.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true but R is NOT the correct explanation of **A.**

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A.**

Q. 159. Which of the following is not a cloning vector?

(1) YAC (2) pBR322

(3) Probe (4) BAC

Q. 160. Given below are two statements:

Statement I: A protein is imagined as a line, the left end represented by first amino acid (C-terminal) and the right end represented by last amino acid (N-terminal).

Statement II: Adult human haemoglobin,consist of 4 subunits (two subunits of α type and two subunits of β type).

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statement I** and **Statement II** are false.

(2) **Statement I** is true but **Statement II** is false.

(3) **Statement I** is false but **Statement II** is true.

(4) Both **Statement I** and **Statement II** are true.

Q. 161. Once the undigested and unabsorb substances center the caecum, their back is prevented by-

(1) lleo-caecal valve

(2) Gastro-oesophageal sphincter

(3) Pyloric sphincter

(4) Sphincter of Oddi

Q. 162. Match **List I** with **List II.**

	List I		List II
A.	Vasectomy	I.	Oral method
B.	Coitus	II.	Barrier method
X.	Cervical caps	III.	Surgical method
Δ.	Saheli	IV.	Natural method

Choose the **correct** answer from the option given below:

(1) A-III, B-IV, C-II, D-I

(2) A-II, B-III, C-I, D-IV

(3) A-IV, B-II,C-I, D-III

(4) A-III, B-I, C-IV, D-II

Q. 163. Which of the following functions is carried out by cytoskeleton in a cell?

(1) Protein synthesis

(2) Motility
(3) Transportation
(4) Nuclear division

Q. 164. Which one of the following common sexually transmitted diseases is completely curable when detected early and treated properly?
(1) Gonorrhoea (2) Hepatitis-B
(3) HIV Infection (4) Genital herpes

Q. 165. Vital capacity of lung is_____.
(1) IRV + ERV+TV+ RV
(2) IRV + ERV + TV – RV
(3) IRV + ERV+TV
(4) IRV + ERV

Q. 166. Which of the following are NOT considered as the part of endomembrane system?
A. Mitochondria
B. Endoplasmic Reticulum
C. Chloroplasts
D. Golgi complex
Choose the **most appropriate** answer from the options below:
(1) A, C and E only
(2) A and D only
(3) A,D and E only
(4) B and D only

Q. 167. Match **List I** with **List II.**

List I		List II	
A.	CCK	I.	Kidney
B.	GIP	II.	Heart
C.	ANF	III.	Gastric gland
D.	ADH	IV.	Pancrease

Choose the **correct** answer from the options given below:
(1) A-III, B-II, C-IV, D-I
(2) A-II, B-IV, C-I, D-III
(3) A-IV, B-II, C-III, D-I
(4) A-IV, B-III, C-II, D-I

Q. 168. Match **List I** with **List II.**

List I		List II	
A.	Gene '*a*'	I.	β-galactosidase
B.	Gene '*y*'	II.	Transacetylase
C.	Gene '*i*'	III.	Permease
D.	Gene '*z*'	IV.	Repressor protein

Choose the **correct** answer from the options given below:
(1) A-II, B-III, C-IV, D-I
(2) A-III, B-IV, C-I, D-II
(3) A-III, B-I, C-IV, D-II
(4) A-II, B-I, C-IV, D-III

. 169. Match **List I** with **List II** with respect to human eye.

	List I		List II
A.	Fovea	I.	Visible coloured portion of eye that regulates diameter of pupil.
B.	Iris	II.	External layer of eye formed of dense connective tissue.
C.	Blind spot	III.	Point of greatest visual acuity or resolution.
D.	Sclera	IV.	Point where optic nerve leaves the eyeball and photoreceptor cells are absent.

Choose the **correct** answer from the options given below:
(1) A-IV, B-III, C-II, D-I
(2) A-I, B-IV, C-III, D-II
(3) A-II, B-I, C-III, D-IV
(4) A-III, B-I, C-IV, D-II

Q. 170. Select the correct group/set of Australian Marsupials exhibiting adaptive radiation.
(1) Numbat, Spotted cuscus, Flying phalanger
(2) Mole, Flying squirrel, Tasmanian tiger cat
(3) Lemur, Anteater, Wolf
(4) Tasmanian wolf, Bobcat, Marsupial mole

Q. 171. Given below are two statements:
Statements I: Vas deferens receives a duct from seminal vesicle and opens into urethra as the ejaculatory duct.
Statements II: The cavity of the cervix is called cervical canal which along with vagina forms birth canal.
In the light of the above statements, choose the **correct** answer from the options given below:
(1) Both **Statements I** and **Statements II** are false.
(2) **Statements I** is correct but **Statements II** is false.
(3) **Statements I** incorrect but **Statement II** is true.
(4) Both **Statement I** and **Statement II** are true.

Q. 172. Match **List I** with **List II.**

	List I		List II
A.	Heroin	I.	Effect on cardiovascular system
B.	Morphine	II.	Slow down body function
C.	Cocaine	II.	Painkiller
D.	Morphine	III.	Interfere with transport of dopamine

Choose the **correct** answer from the options given below:
(1) A-I, B-II, C-III, D-IV
(2) A-IV, B-III, C-II, D-I
(3) A-III, B-IV, C-I, D-II
(4) A-II, B-I, C-IV, D-III

Q. 173. Given below are two statements:

	List I		List II
A.	Ringworm	I.	*Haemophilus influenzae*
B.	Filariasis	II.	*Trichophyton*
X.	Malaria	III.	*Wuchereria bancrofti*
Δ.	Pneumonia	IV.	*Plasmodium vivax*

Choose the **correct** answer from the options given below:

(1) A-II, B-III, C-I, D-IV
(2) A-III, B-II, C-I, D-IV
(3) A-III,B-II, C-IV, D-I
(4) A-II, B-II, C-IV, D-I

Q. 174. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.
Assertion A: Amniocentesis for sex determination is one of the strategies of Reproductive and Child Health Care Programme.
Reason R: Ban on amniocentesis checks increasing menace of female foeticide.
In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true and **R** is NOT the correct explanation of **A**.
(2) **A** is true but **R** is false.
(3) **A** is false but **R** is true.
(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 175. Radial symmetry is not found in adults of phylum_____.

(1) Hemichordate **(2)** Coelenterata
(3) Echinodermata **(3)** Ctenophora

Q. 176. Given below are two statements:
Statement I: Low temperature preserves the enzyme in a temporarily inactive state activity because proteins are denatured by whereas high temperature destroys enzymatic activity because proteins are denatured by heat.
Statement II: When the inhibitor closely resembles the substrate in its molecular structure and inhibits the activity of the enzyme, it is known as competitive inhibitor.
In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **Statements I** and **Statements II** are false.
(2) **Statements I** is true but **Statements II** is false.
(3) **Statements I** is false but **statements II** is true.
(4) Both **Statements I** and **statements II** are true.

Q. 177. Match **List I** and **List II.**

	List I		List II
A.	Taenia	I.	Nephridia
B.	*Paramoecium*	II.	Contractile vacuole
C.	*Periplaneta*	III.	Flame cells
D.	*Pheretima*	IV.	Urecose gland

Choose the **correct** answer from the options give below:

(1) A-1, B-II, C-IV, D-III
(2) A-III, B-II, C-IV, D-I
(3) A-II, B-I, C-IV, D-III
(4) A-I, B-II, C-III, D-IV

Q. 178. Which of the following statements are correct regarding female reproductive cycle?

A. In non-primate mammals cyclical changes during reproduction are called oestrus cycle.
B. First menstrual cycle begins at puberty and is called menopause.
C. Lack of menstruation may be indicative of pregnancy.
D. Cyclic menstruation extends between menarche and menopause.

Choose the **most appropriate** answer from the options given below:

(1) A and B only
(2) A, B and C only
(3) A, C and D only
(4) A and D only

Q. 179. Given below are two statements:
Statements I: Electrostatic precipitator is most widely used in thermal power plant.
Statement II: Electrostatic precipitator in thermal power plant removes ionising radiations
In the light of the above statements, choose the **most appropriate** answer from the options given below:

(1) Both **Statement I** and **Statement II** are incorrect.
(2) **Statement I** is correct but **Statement II** is incorrect.
(3) **Statement I** incorrect but **Statement II** is correct.
(4) Both **Statement I** and **Statement II** are correct.

Q. 180. Match List I and List II.

	List I (Cells)		List II (Secretion)
A.	Peptic cells	I.	Mucus
B.	Goblet cells	II.	Bile juice
C.	Oxyntic cells	III.	Proenzyme pepsinogen
D.	Hepatic cells	IV.	HCl and intrinsic factor for absorption of vitamin B_{12}

Choose the **correct** answer from the options give below:

(1) A-II, B-I, C-III, D-IV
(2) A-III, B-I, C-IV, D-II
(3) A-II, B-IV, C-I, D-III
(4) A-IV, B-III, C-II, D-I

Q. 181. Given below are two statements: one labelled as **Assertion A** and the other is labelled as **Reason R**.
Assertion A: Endometrium is necessary for implantation of blastocyst.

Reason R: In the absence of fertilization, the corpus luteum degenerates that causes disintegration of endometrium.

In the light of the above statements, choose the **correct** answer from the options given below:

(1) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.

(2) **A** is true but **R** is false.

(3) **A** is false but **R** is true.

(4) Both **A** and **R** are true and **R** is the correct explanation of **A**.

Q. 182. Match **List I** with **List II**.

List (Type of Joint)		List II (Found between)	
A.	Cartilaginous	I.	Between flat skull bones
B.	Ball and Socket Joint	II.	Between adjacent vertebrate in vertebral column
X.	Fibrous Joint	III.	Between carpal and metacarpal of thumb
Δ.	Saddle Joint	IV.	Between humerus and pectoral girdle

Choose the **correct** answer from the options given below:

(1) A-II, B-IV, C-I, D-III

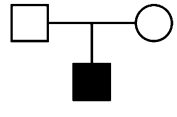
(2) A-I, B-IV, C-III, D-II

(3) A-II, B-IV, C-III, D-I

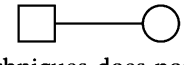
(4) A-III, B-I, C-II, D-IV

Q. 183. Which one of the following symbols represents mating between relatives in human pedigree analysis?

(1) 

(2) 

(3) 

(4) 

Q. 184. Which one of the following techniques does not serve the purpose of early diagnosis of a disease for its early treatment?

(1) Serum and Urine analysis

(2) Polymerase Chain Reaction (PCR) technique

(3) Enzyme Linked Immuno-Sorbent Assay (ELISA) technique

(4) Recombinant DNA Technology

Q. 185. Match **List I** with **List II**.

List I (Interacting species)		List II (Name of interaction)	
A.	A leopard and a lion in a forest/ grassland	I.	Competition
B.	A cuckoo laying egg in a crow's nest	II.	Brood parasitism
C.	Fungi and root of a higher plant in mycorrhizae	III.	Mutualism
D.	A cattle egret and a cattle in a field	IV.	Commensalism

Choose the **correct** answer from the options given below:

(1) A-I, B-II, C-IV, D-III

(2) A-III, B-IV, C-I, D-II

(3) A-II, B-III, C-I, D-IV

(4) A-I, B-II, C-III, D-IV

[redacted]

Q. 186. Given below are two statements:

Statement I: During G_0 phase of cell cycle, the cell is metabolically inactive.

Statement II: The centrosome undergoes duplication during S phase of interphase.

In the light of the above statements, choose the **most appropriate** answer from the options given below:

(1) Both **Statement I** and **Statement II** are incorrect.

(2) **Statement I** is correct but

(3) **Statement I** is incorrect but **Statement II** is correct.

(4) Both **Statement I** and **Statement II** are correct.

Q. 187. Match **List I** with **List II**.

List I		List II	
A.	Logistic growth	I.	Unlimited resource availability condition
B.	Exponential	II.	Limited resource availability condition
C.	Expanding age pyramid	III.	The percent individuals of pre reproductive age is largest followed by reproductive and post reproductive age groups
D.	Stable age pyramid	IV.	The percent individuals of pre-reproductives and reproductive age group are same

Choose the **correct** answer from the options given below:

(1) A-II, B-III, C-I, D-IV

(2) A-II, B-IV, C-I, D-III

(3) A-II, B-IV, C-III, D-I

(4) A-II, B-I, C-III, D-IV

Q. 188. In cockroach, excretion is brought about by-

A. Phallic gland **B.** Urecose gland

C. Nephrocytes **D.** Fat body

E. Collaterial glands

Choose the **correct** answer from the options given below:

(1) A, B and E only **(2)** B, C and D only

(3) B and D only **(4)** A and E only

Q. 189. Select the correct statements.

A. Tetrad formation is seen during leptotene.

B. During anaphase, the centromeres split and chromatids separate.

C. Terminalization takes place during pachytene.

D. Nucleolus, golgi complex and ER are reformed during telophase.

E. Crossing over takes place between sister chromatids of homologous chromosome.

Choose the **correct** answer from the options given below:

(1) B and D only

(2) A , C and E only

(3) B and E only

(4) A and C only

Q. 190. Which of the following are NOT under the control of thyroid hormone?

A. Maintenance of water and electrolyte balance.

B. Regulation of basal metabolic rate.

C. Normal rhythm of sleep-wake cycle.

D. Development of immune system.

E. Support the process of R.B.Cs formation.

Choose the **correct** answer from the options given below:

(1) B and C only

(2) C and D only

(3) D and E only

(4) A and D only

Q. 191. which of the following is characteristic feature of cockroach regarding sexual dimorphism?

(1) Presence of anal styles

(2) Presence of sclerites

(3) Presence of anal cerci

(4) Dark brown body colour and anal cerci

Q. 192. Which of the following statements are correct?

A. Basophils are most abundant cells of the total WBCs.

B. Basophils secrete histamine, serotonin and heparin.

C. Basophils are involved in inflammatory response.

D. Basophils have kidney shaped nucleus.

E. Basophils are agranulocytes.

Choose the **correct** answer from the options given below:

(1) C and E only **(2)** B and C only

(3) A and B only **(4)** D and E only

Q. 193. Which of the following statements are correct?

A. An excessive loss of body fluid from the body switches off osmoreceptors.

B. ADH facilitates water reabsorption to prevent diuresis.

C. ANF causes vasodilation.

D. ADH causes increase in blood pressure.

E. ADH is responsible for decrease in GFR.

Choose the **correct** answer from the options given below:

(1) B, C and D only

(2) A, B and E only

(3) C, D and E only X

(4) A and B only

Q. 194. Match **List I** with **List II.**

List I		List II	
A.	Mast cells	I.	Ciliated epithelium
B.	Inner surface of bronchiole	II.	Areolar
C.	Blood.	III.	Cuboidal epithelium
D.	Tubular parts	IV.	specialised connective tissue

Choose the **correct** answer from the options give below:

(1) A-II, B-III, C-I, D-IV

(2) A-II, B-I, C-IV, D-III

(3) A-III, B-IV, C-II, D-I

(4) A-I, B-II, C-IV, D-III

Q. 195. Select the correct statements with reference to chordates.

A. Presence of a mid-dorsal, solid and double nerve cord.

B. Presence of closed circulatory system.

C. Presence of paired pharyngeal gillslits.

D. Presence of dorsal heart

E. Triploblastic pseudocoelomate animals.

Choose the **correct** answer from the options given below:

(1) B and C only

(2) B, D and E only

(3) C, D and E only

(4) A, C and D only

Q. 196. The unique mammalian characteristics are:

(1) hairs, pinna and mammary glands

(2) hairs, pinna and indirect development

(3) pinna, monocondylic skull and mammary glands

(4) hairs, tympanic membrane and mammary glands

Q. 197. The parts of human brain that helps in regulation of sexual behaviour, expression of excitement, pleasure, rage, fear etc. are :

(1) Corpora quadrigemina & hippocampus

(2) Brain stem & epithalamus

(3) Corpus callosum and thalamus

(4) Limbic system & hypothalamus

Q. 198. Which one of the following is the sequenate on corresponding coding strand, if the sequence on *mRNA* formed follows
5′AUCGAUCGAUCGAUCGAUGG AUCG AUCG 3′?

(1) 3′ UAGCUAGCUAGCUAGCUA GCUAGCUAGC 5′

(2) 5′ ATCGATCGATCGATCGATCG ATCGATCG 3′

(3) 3′ ATCGATCGATCGATCGATG ATCGATCG 5′

(4) 5′ UAGCUAGCUAGCUAGCUA GCUAGC UAGC 3′

Q. 199. Which of the following statements are correct regarding skeletal muscle?

A. Muscle bundles are held together by collagenous connective tissue layer called fascicle.

B. Sarcoplasmic reticulum of muscle fiber is a store house of calcium ions.

C. Striated appearance of skeletal muscle fibre is due to distribution pattern actin and myosin proteins.

D. M line is considered as functional of contraction called sarcomere.

Choose the **most appropriate** answer from the options given below:

(1) B and C only

(2) A, C and D only

(3) C and D only

(4) A, B and C only

Q. 200. Which one of the following is NOT an advantage of inbreeding?

(1) It exposes harmful recessive genes that are eliminated by selection.

(2) Elimination of less desirable genes and accumulation of superior genes takes place due to it.

(3) It decreases the productivity of inbred population, after continuous inbreeding.

(4) It decreases homozygosity.

■■■